










<p>Scientific Models of the Atom</p>  <p>What ideas did the Ancient Greeks have about the atom?</p>	<p>mrbakerssciencestuff.com</p> <p>1 of 10</p>
<p>Scientific Models of the Atom</p>  <p>Describe John Dalton's Atomic Model.</p>	<p>mrbakerssciencestuff.com</p> <p>2 of 10</p>
<p>Scientific Models of the Atom</p>  <p>Describe J.J. Thompson's Atomic Model.</p> <p>Draw a diagram</p>	<p>mrbakerssciencestuff.com</p> <p>3 of 10</p>
<p>Scientific Models of the Atom</p>  <p>Describe Rutherford's Atomic Model.</p> <p>Draw a diagram</p>	<p>mrbakerssciencestuff.com</p> <p>4 of 10</p>

Instructions:

- (1) Answer the questions.
- (2) Watch the clip and correct your answers.
- (3) Print out, fold over on dotted line and make into flashcards.
- (4) Use for retrieval quizzes.





<p>Scientific Models of the Atom</p>  <p>Describe Rutherford's Gold Leaf Experiment.</p>	<p>mrbakerssciencestuff.com</p> <p>5 of 10</p>
<p>Scientific Models of the Atom</p>  <p>What were the observations made during Rutherford's Gold Leaf Experiment?</p>	<p>mrbakerssciencestuff.com</p> <p>6 of 10</p>
<p>Scientific Models of the Atom</p>  <p>What conclusions about atoms and their nuclei did Rutherford make?</p>	<p>mrbakerssciencestuff.com</p> <p>7 of 10</p>
<p>Scientific Models of the Atom</p>  <p>Describe Bohr's Model of the Atom.</p> <p>Draw a diagram.</p>	<p>mrbakerssciencestuff.com</p> <p>8 of 10</p>

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