




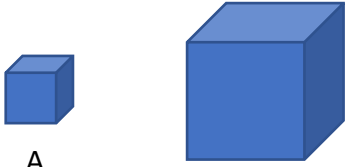







| | |
|---|--|
| <p>Density</p>  <p>Draw particle diagrams of a solid, liquid and a gas.</p> | <p>mrbakerssciencestuff.com</p> <p>1 of 11</p> |
| <p>Density</p>  <p>Define Density.</p> | <p>mrbakerssciencestuff.com</p> <p>2 of 11</p> |
| <p>Density</p>  <p>Explain why a can of normal coke sinks in water but a can of diet can floats.</p>  <p><small>© Gphar®, 2003</small></p> | <p>mrbakerssciencestuff.com</p> <p>3 of 11</p> |
| <p>Density</p>  <p>A student has 2 objects, labelled A and B below. they have the same mass. Which one is the densest?</p>  <p>A B</p> | <p>mrbakerssciencestuff.com</p> <p>4 of 11</p> |

Instructions:

- (1) Answer the questions.
- (2) Watch the clip and correct your answers.
- (3) Print out, fold over on dotted line and make into flashcards.
- (4) Use for retrieval quizzes.





| | |
|---|--|
| <p>Density</p>  <p>How many ml in a litre?</p> | <p>mrbakerssciencesstuff.com</p> <p>9 of 11</p> |
| <p>Density</p>  <p>How many cm³ in a litre?</p> | <p>mrbakerssciencesstuff.com</p> <p>10 of 11</p> |
| <p>Density</p>  <p>How many cm³ in a m³?</p> | <p>mrbakerssciencesstuff.com</p> <p>11 of 11</p> |

Instructions:

- (1) Answer the questions.
- (2) Watch the clip and correct your answers.
- (3) Print out, fold over on dotted line and make into flashcards.
- (4) Use for retrieval quizzes.

